

Australian Government Department of Industry,

#### Department of Industry, Science and Resources

### National Measurement Institute



# REFERENCE MATERIAL PRODUCT INFORMATION SHEET

### NMIA D595a: 19-Noretiocholanolone glucuronic acid sodium salt

**IUPAC name:** Sodium  $(3\alpha,5\beta)$ -17-Oxoestran-3-yl  $\beta$ -D-glucopyranosiduronate

**Expiration of certification:** The property values are valid till 27/07/2033, i.e. ten years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to ampoules that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** The compound is supplied as a dried aliquot in a sealed ampoule under an atmosphere of argon. The RM is intended for a single use to prepare a standard solution containing D595a. Material was prepared by synthesis, and certified for identity and purity by NMIA.

Intended use: This reference material should be used for qualitative analysis only.

**Instructions for use:** Open the ampoule and carefully rinse the interior at least three times with a suitable organic solvent (e.g. methanol). This will transfer 816  $\mu$ g of anhydrous 19-noretiocholanolone glucuronide. The mass of analyte in each ampoule is calculated from the assigned purity of the bulk and the concentration of bulk material in a stock solution used to prepare the ampoules.

Recommended storage: When not in use, this material should be stored at or below 4 °C in a closed container in a dry, dark area.

**Stability:** This material has demonstrated stability over a minimum period of five years. The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by HPLC with ELS detection on seven randomly selected ampoules of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Safety:** Treat as hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 4 August 2023

This report supersedes any issued prior to 4 August 2023.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

HPLC:	Instrument:	Waters Model 1525 Binary pump, 717 plus autosampler OR Thermo Ultimate 3000	
	Column:	Alltima C-18, 5 μm (4.6 mm x 150 mm)	
	Mobile Phase:	A = 20 mM ammonium acetate buffer (pH 4.2); B = Acetonitrile	
		0-8 min 28% B; 8-9 min 28-52% B; 9-12 min 52% B	
	Flow Rate:	1.0 mL/min	
	Detector:	Waters ELSD 2424 or RS CAD Ultra	
	Relative peak area of the main component:		
	Initial analysis:	Mean = 99.9%, s = 0.03% (7 ampoules in duplicate, July 2010)	
	Re-analysis:	Mean = 100.0%, s = 0.0% (5 ampoules in duplicate, August 2011)	
	Re-analysis:	Mean = 100.0%, s = 0.02% (5 ampoules in duplicate, July 2014)	
	Re-analysis:	Mean = $100.0\%$ , s = $0.0\%$ (5 ampoules in duplicate, June 2019)	
	Re-analysis:	Mean = 100.0%, s = 0.0% (5 ampoules in duplicate, July 2023)	

#### The following analytical data was obtained on the bulk material subsequently used in the preparation of the ampoules.

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The purity value by qNMR was obtained using a combination of the three-proton singlet at 0.79 ppm and the one-proton multiplet at 2.9 ppm measured against a certified internal standard of dimethyl terephthalate.

Supporting evidence is provided by HPLC with ELS detection, thermogravimetric analysis, Karl Fischer analysis, <sup>1</sup>H NMR spectroscopy, qualitative headspace GC-MS analysis of occluded solvents and elemental microanalysis.

QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis:	Bruker DMX-400 400 MHz DMSO- $d_6$ Dimethyl terephthalate (100% mass fraction) Mean = 81.0%, s = 0.7% (3 sub samples in duplicate, September 2011)	
HPLC:	Instrument: Column: Mobile Phase: Flow Rate: Detector:	Waters Model 1525 Binary pump, 717 plus autosampler Alltima C-18, 5 $\mu$ m (4.6 mm × 150 mm) Acetonitrile/20 mM ammonium acetate (pH 4.2) [25:75] 1.0 mL/min Waters ELSD 2424	
	Relative peak area of the main component: Initial analysis: Mean = $99.9\%$ , s = $0.02\%$ (7 sub samples in duplicate, July 2010)		
Thermogravimetric analysis:		Volatiles content 12.1% mass fraction. Non-volatile residue was not determined (November 2007)	
Karl Fischer analysis:		Moisture content 12.3% mass fraction (November 2007) Moisture content 12.2% mass fraction (March 2010)	

## Spectroscopic and other characterisation data

GC-MS:	Persilylated derivative: Instrument: Column: Program: Injector: Transfer line temp: Carrier: Split ratio:	HP6890/5973 HP Ultra 1, 17 m × 0.22 mm l.D. × 0.11 μm 200 °C, 10 °C/min to 300 °C (3 min) 280 °C 300 °C Helium, 1.0 mL/min 15/1	
	The retention time of the <i>persilylated</i> derivative is reported along with the major peaks in the mass spectrum. The latter are reported as mass/charge ratios and (in brackets) as a percentage relative to the base peak. The molecular ion is not observed.		
	<i>Per</i> -TMS (11.5 min):	490 (7), 331 (39), 305 (15), 292 (16), 217 (66), 204 (46), 73 (100) <i>m/z</i>	
ESI-MS:	Instrument: Operation: Ionisation: Peak:	Finnigan MAT TSQ 700 Negative ion mode, direct infusion ESI probe at 4.5 kV 451.2 <i>m/z</i> (M-Na) <sup>-</sup>	
FAB-MS:	lons: Ionisation:	497(M-Na)⁺, 475 (MH)⁺, 439, 413, 329, 286 <i>m/z</i> 15 kVin NBA/MeOH	
HRMS:	Found <i>m/z</i> 497.209; C <sub>24</sub> H <sub>35</sub> O <sub>8</sub> Na <sub>2</sub> (MNa <sup>+</sup> ) requires <i>m/z</i> 497.213 <i>m/z</i>		
IR:	Instrument: Range: Peaks:	Perkin-Elmer FT-IR 4000-400 cm <sup>-1</sup> , Nujol mull 3488, 1733, 1609, 1454, 1376, 1159, 1078, 1026 cm <sup>-1</sup>	
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Key spectral data:	Bruker DMX-500 500 MHz MeOH- $d_4$ (3.31 ppm) $\delta$ 0.91 (3H, s), 3.20 (1H, dd), 3.41 (1H, t), 3.44 (1H, dd), 3.58 (1H, d), 3.84 (1H, m), 4.45 (1H, d) ppm Ethanol was observed in the <sup>1</sup> H NMR at 1.5% mass fraction (May 2000, December 2007 and September 2011)	
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker DMX-500 126 MHz MeOH- $d_4$ (49.0 ppm) $\delta$ 14.2, 22.6, 26.0, 26.2, 26.7, 26.9, 32.5, 32.9, 35.1, 36.8, 37.2, 40.0, 41.6, 42.6, 51.9, 73.8, 75.0, 76.2, 77.9, 79.3, 101.9, 177.0, 228 ppm One signal obscured by solvent peak	
Microanalysis:	Found: Calculated:	C = 52.6%; H = 7.4% (September 2010) C = 60.8%; H = 7.4% (Calculated for $C_{24}H_{35}O_8Na$ )	