

Australian Government Department of Industry,

Science and Resources

## National Measurement Institute



# CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

## NMIA D958: (±)-N,N-Dimethylcathinone hydrochloride

Report ID: D958.2022.01

Chemical Formula: C11H16CINO

Molecular Weight: 213.7 g/mol (HCl), 177.2 g/mol (base)

### **Certified value**

Batch No.	CAS No.	Purity (mass fraction)
10-D-11	10105-90-5 (HCI) 15351-09-4 (base)	99.4 ± 0.5%

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit (k = 2).

IUPAC name: 2-(Dimethylamino)-1-phenyl-1-propanone hydrochloride

**Expiration of certification:** The property values are valid till 5 July 2027, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** White crystalline solid prepared by synthesis, certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

Intended use: This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

Recommended storage: When not in use this material should be stored at or below 25 °C in a closed container in a dry, dark area.

**Metrological traceability:** The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. In the mass balance approach all impurities are quantified as a mass fraction and subtracted from 100%. Quantitative NMR provides an independent direct measure of the mass fraction of the analyte of interest, calibrated with an internal standard certified for purity (mass fraction).

**Stability:** This material has demonstrated stability over a minimum period of three years. The measurement uncertainty at the 95% confidence interval includes a stability component which has been estimated from annual stability trials. The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by HPLC with UV detection on ten randomly selected 1-2 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Caution:** Treat as a hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

Report ID: D958.2022.01 Product release date: 12 January 2011

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 12 July 2022

This report supersedes any issued prior to 12 July 2022.

NATA accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The certified purity value was obtained from a combination of traditional analytical techniques including GC-FID, thermogravimetric analysis, Karl Fischer analysis and <sup>1</sup>H NMR spectroscopy. The purity value is calculated as per Equation 1. Supporting evidence is provided by HPLC with UV detection and quantitative nuclear magnetic resonance (qNMR).

Purity =  $(100 \% - I_{ORG}) \times (100 \% - I_{VOL} - I_{NVR})$ 

Equation 1

I<sub>ORG</sub> = Organic impurities of related structure, I<sub>VOL</sub> = volatile impurities, I<sub>NVR</sub> = non-volatile residue.

Supporting evidence is provided by qualitative headspace GC-MS analysis of occluded solvents and elemental microanalysis.

GC-FID:	Instrument: Column: Program: Injector: Detector Temp: Carrier: Split ratio:	Agilent 6890N or 8890 HP-1, 30 m × 0.32 mm l.D. × 0.25 μm 100 °C (1 min), 8 °C/min to 160 °C, 30 °C/min to 300 °C (3 min) 250 °C 320 °C Helium 20/1
	Relative mass fractiono Initial analysis: Re-analysis: Re-analysis: Re-analysis:	f the main component as the free base: Mean = 99.9%, s = 0.003% (5 sub samples in duplicate, October 2013) Mean = 99.6%, s = 0.01% (5 sub samples in duplicate, August 2016) Mean = 99.7%, s = 0.11% (5 sub samples in duplicate, July 2017) Mean = 99.8%, s = 0.01% (5 sub samples in duplicate, July 2022)
HPLC:	Instrument:	Waters Model 1525 Binary pump, 717 plus Autosampler or Shimadzu Model LC-20AB Binary pump, SIL-20A HT autosampler
	Column:	Alltima C-18, 5 μm (4.6 mm x 150 mm)
	Column oven:	40 °C
	Mobile Phase:	Acetonitrile/MilliQ Water (50:50 v/v) The aqueous phase was buffered at pH 10.8 using 20mM NH₄OAc and NH₃
	Flow rate:	1.0 mL/min
	Detector:	Waters 2998 PDA operating at 242 nm or
	Delector.	Shimadzu PDA SPD-M20A operating at 242 nm
	Relative mass fraction of the main component:	
	Initial analysis:	Mean = 99.95%, s = 0.003% (10 sub samples in duplicate, October 2010)
	Re-analysis:	Mean = 99.99%, s = 0.002% (5 sub samples in duplicate, October 2011)
	Re-analysis:	Mean = 99.95%, s = 0.01% (5 sub samples in duplicate, November 2012)
Karl Fischer analysis:		Moisture content < 0.2% mass fraction (December 2010, November 2011, November 2012 and October 2013) Moisture content 0.6% mass fraction (August 2016) Moisture content 0.3% mass fraction (July 2017 & July 2022)
Thermogravimetric analysis:		Non volatile residue < 0.2 % mass fraction (November 2010). The volatile content (e.g. organic solvents and/or water) could not be analysed accurately because of the inherent volatility of the material.
QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis:	Bruker Avance-400 400 MHz $D_2O$ (4.79 ppm) Maleic acid (98.7% mass fraction) Mean = 99.7%, s = 0.49% (5 sub samples, December 2010)

### Spectroscopic and other characterisation data

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GC-MS:	Instrument: Column:	Agilent 6890/5973 TG-1MS, 30 m x 0.25 mm I.D. x 0.25 μm
	Program:	60 °C (1 min), 10 °C/min to 100 °C, 15 °C/min to 250 °C (5 min), 30 °C/min to 300 °C (2 min)
	Injector: Transfer line temp: Carrier: Split ratio:	250 °C 300 °C Helium, 1.0 mL/min 20/1
		e free base is reported with the major peaks in the mass spectra. The latter are reported and (in brackets) as a percentage relative to the base peak. 175 (1), 146 (3), 105 (4), 77 (9), 72 (100), 42 (6) $m/z$
HS-GC-MS:	Instrument: Column: Program: Injector: Transfer line temp: Carrier: Split ratio: Solvents detected:	Agilent 6890/5973/G1888 DB-624, 30 m x 0.25 mm I.D. x 1.4 μm 50 °C (5 min), 7 °C/min to 120 °C, 15 °C/min to 220 °C (8.3 min) 150 °C 280 °C Helium, 1.2 mL/min 50/1 Ethanol
TLC:	Conditions:	Kieselgel 60F <sub>254</sub> . Methanol/NH <sub>3</sub> (100/1.5) Single spot observed, $R_f = 0.61$ . Visualisation with UV at 254 nm
IR:	Instrument: Range: Peaks:	Biorad FTS300MX FT-IR 4000-400 cm <sup>-1</sup> , KBr powder 3357, 2950, 2763, 2634, 2432, 1997, 1855, 1689, 1594, 1472, 1453, 1370, 1302, 1236, 1155, 1109, 1065, 1022, 968, 922, 798, 710, 665, 475 cm <sup>-1</sup>
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance III-400 400 MHz $D_2O$ (4.79 ppm) $\delta$ 1.63 (3H, d, $J$ = 7.3 Hz), 3.00 (6H, s), 5.25 (1H, q, $J$ = 7.3 Hz), 7.63 (2H, m), 7.79 (1H, m), 8.02 (2H, m) ppm Ethanol estimated at 0.08% mass fraction was observed in the <sup>1</sup> H NMR
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance III-400 100 MHz D₂O δ 14.0, 40.0, 43.0, 66.1, 128.9, 129.3, 132.5, 135.6, 197.4 ppm
Melting point:		195-199 °C
Microanalysis:	Found: Calculated:	C = 61.8%; H = 7.6%; N = 6.6%; Cl = 16.6% (December, 2010) C = 62.0%; H = 7.8%; N = 6.6%; Cl = 16.5% (Calculated for $C_{11}H_{16}CINO$ )