

Australian Government

Department of Industry, Science and Resources

## National Measurement Institute



# CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

## NMIA D957: (±)-N,N-Diethylcathinone hydrochloride

Report ID: D957.2022.02 (Bottled 160823)

Chemical Formula: C13H20CINO

Molecular Weight: 241.8 g/mol

### **Certified value**

Batch No.	CAS No.	Purity (mass fraction)
10-D-10	134-80-5	97.2 ± 0.9%

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit (k = 2).

IUPAC: 2-(Diethylamino)-1-phenyl-1-propanone hydrochloride

**Expiration of certification:** The property values are valid till 9 February 2027, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** Off-white powder prepared by synthesis, certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

Intended use: This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

Recommended storage: When not in use this material should be stored at or below 4 °C in a closed container in a dry, dark area.

**Metrological traceability:** The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. In the mass balance approach all impurities are quantified as a mass fraction and subtracted from 100%. Quantitative NMR provides an independent direct measure of the mass fraction of the analyte of interest, calibrated with an internal standard certified for purity (mass fraction).

**Stability:** This material has demonstrated stability over a minimum period of five years. The measurement uncertainty at the 95% coverage interval includes a stability component which has been estimated from long term stability trials.

The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by HPLC with UV detection on ten randomly selected 1-2 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Caution:** Treat as a hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

Report ID: D957.2022.02 (Bottled 160823) Product release date: 23 August 2010

measurement.gov.au

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 20 September 2022

This report supersedes any issued prior to 20 September 2022.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The certified purity value was obtained by mass balance from a combination of traditional analytical techniques, including GC-FID, HPLC with UV detection, qNMR, thermogravimetric analysis, Karl Fischer analysis and <sup>1</sup>H NMR spectroscopy. The purity value is calculated as per Equation 1.

Purity = (100 % -  $I_{ORG}$ ) x (100 % -  $I_{VOL} - I_{NVR}$ )

Equation 1

 $I_{ORG}$  = Organic impurities of related structure,  $I_{VOL}$  = volatile impurities,  $I_{NVR}$  = non-volatile residue.

Supporting evidence is provided by qualitative headspace GC-MS analysis of occluded solvents and elemental microanalysis.

GC-FID:	Instrument: Column: Program: Injector: Detector Temp: Carrier: Split ratio:	Agilent 7890A/6890 HP-5, 30 m × 0.32 mm l.D. × 0.25 μm 100 °C (1 min), 10 °C/min to 180 °C, 30 °C/min to 300 °C (3 min) 250 °C 320 °C Helium 20/1
	Relative mass fraction Initial analysis: Re-analysis: Re-analysis: Re-analysis: Re-analysis:	of main component as the free base: Mean = 99.5%, s = 0.08% (5 sub samples in duplicate, August 2012) Mean = 99.7%, s = 0.04% (5 sub samples in duplicate, June 2013) Mean = 99.7%, s = 0.03% (5 sub samples in duplicate, April 2016) Mean = 99.7%, s = 0.01% (5 sub samples in duplicate, March 2019) Mean = 99.6%, s = 0.02% (5 sub samples in duplicate, February 2022)
HPLC:	Instrument: Column: Mobile Phase: Flow rate: Detector:	Shimadzu Model LC-20AB Binary pump, SIL-20A HT autosampler or Waters Binary pump 1525, 717 plus autosampler X-Bridge C-18, 5 μm (4.6 mm x 150 mm) (40 °C) Acetonitrile/Milli-Q water NH <sub>4</sub> OAc buffer (60:40) The aqueous phase was buffered at pH 10.8 NH <sub>4</sub> OAc 1.0 mL/min Shimadzu PDASPD-M20A or Waters PDA-2998 operating at 240 nm
	Relative mass fraction Initial analysis: Re-analysis:	of main component: Mean = 99.9%, s = 0.01% (10 sub samples in duplicate, August 2010) Mean = 99.9%, s = 0.01% (5 sub samples in duplicate, August 2011)
Thermogravimetric analysis:		Non volatile residue < 0.2% mass fraction (November 2010). The volatile content (e.g. organic solvents and/or water) could not be analysed accurately because of the inherent volatility of the material.
Karl Fischer ana	alysis:	Moisture content 0.3% mass fraction (September 2010) Moisture content 0.6% mass fraction (June 2013) Moisture content 0.8% mass fraction (May 2014) Moisture content 1.1% mass fraction (March 2019) Moisture content 1.0% mass fraction (January 2022)
QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis:	Bruker Avance-400 400 MHz D <sub>2</sub> O (4.79 ppm) Potassium hydrogen maleate (98.8% mass fraction) Mean = 97.3%, s = 0.25% (5 sub samples, August 2010)

### Spectroscopic and other characterisation data

LC/ESI -MS:	Instrument: Operation: Ionisation: EM voltage: Cone voltage: Peak:	Micromass Quatro LC Micro Positive ion mode, direct infusion at 10 μL/min ESI spray voltage at 3.5 kV positive ion 650 V 20 V 206.1 (M+H <sup>+</sup> ) <i>m/z</i>		
HS-GC-MS:	Instrument: Column: Program: Injector: Transfer line temp: Carrier: Split ratio: Solvents detected:	Agilent 6890/5973/G1888 DB-624, 30 m x 0.25 mm l.D. x 1.4 μm 50 °C (5 min), 7 °C/min to 120 °C, 15 °C/min to 220 °C (8.3 min) 150 °C 280 °C Helium, 1.2 mL/min 50/1 Ethanol, diethyl ether and ethyl acetate		
TLC:	Conditions:	Kieselgel 60F254. Methanol/NH3 (100/1.5) Single spot observed, Rf = 0.7. Visualisation with UV at 254 nm		
IR:	Instrument: Range: Peaks:	Biorad FTS300MX FT-IR 4000-400 cm <sup>-1</sup> , KBr powder 3353, 2983, 2943, 2616, 1687, 1447, 1390, 1290, 1233, 1129, 1015, 976, 707 cm <sup>-1</sup>		
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance DMX-600 600 MHz $D_2O$ (4.79 ppm) $\delta$ 1.37 (3H, t, $J$ = 7.0 Hz), 1.45 (3H, t, $J$ = 7.0 Hz), 1.64 (3H, d, $J$ = 7.1 Hz), 3.20 (1H, m), 3.37 (2H, m), 3.54 (1H, m), 5.30 (1H, q, $J$ = 7.3 Hz), 7.65 (2H, t, $J$ = 7.8 Hz), 7.82 (2H, t, J = 7.5 Hz), 8.07 (2H, t, $J$ = 7.5 Hz) ppm		
		Ethanol and diethylamine hydrochloride estimated at 0.4% and 1.1% mass fraction respectively were observed in the <sup>1</sup> H NMR.		
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance DMX-600 151 MHz D <sub>2</sub> O δ 7.0, 9.1, 13.7, 44.0, 47.2, 62.2, 128.5, 128.9, 132.2, 135.2, 197.1 ppm		
Melting point:		154-164 °C		
Microanalysis:	Found: Calculated:	C = 64.2%; H = 8.5%; N = 5.9%; CI = 14.8% (August, 2010) C = 64.6%; H = 8.3%; N = 5.8%; CI = 14.7% (Calculated for $C_{13}H_{20}CINO$ )		