

Australian Government

Department of Industry, Science and Resources

## National Measurement Institute



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HCI

# CERTIFIED REFERENCE MATERIAL CERTIFICATE OF ANALYSIS

# NMIA D775b: (+)-Pseudoephedrine hydrochloride

Report ID: D775b.2021.02 (Bottled 201117)

Chemical Formula: C<sub>10</sub>H<sub>15</sub>NO.HCl

Molecular Weight: 201.7 g/mol (HCl), 165.2 g/mol (base)

### **Certified value**

Batch No.	CAS No.	Purity (mass fraction)
12-D-04	345-78-8 (HCI) 90-82-4 (base)	99.9 ± 0.4%

The uncertainty has been calculated according to ISO Guide 35 and is stated at the 95% confidence limit (k = 2).

IUPAC name: 2-(Methylamino)-1-phenyl-1-propanol hydrochloride

**Expiration of certification:** The property values are valid till 12 August 2026, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to sample bottles that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** Off-white powder sourced from an external supplier, certified for identity and purity by NMIA. Packaged in amber glass bottles with a septum and crimped aluminium cap or screw top cap.

Intended use: This certified reference material is suitable for use as a primary calibrator.

Instructions for use: Equilibrate the bottled material to room temperature before opening.

Recommended storage: When not in use this material should be stored at or below 25 °C in a closed container in a dry, dark area.

**Metrological traceability:** The certified purity value is traceable to the SI unit for mass (kg) through Australian national standards via balance calibration. In the mass balance approach all impurities are quantified as a mass fraction and subtracted from 100%. Quantitative NMR provides an independent direct measure of the mass fraction of the analyte of interest, calibrated with an internal standard certified for purity (mass fraction).

**Stability:** This material has demonstrated stability over a minimum period of three years. The measurement uncertainty at the 95% confidence interval includes a stability component which has been estimated from annual stability trials. The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by GC-FID on ten randomly selected 1-2 mg sub samples of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Caution:** Treat as a hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 15 September 2022

This report supersedes any issued prior to 15 September 2022.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

The purity value was obtained from a combination of traditional analytical techniques and quantitative nuclear magnetic resonance (qNMR). The techniques used in the mass balance approach include GC-FID thermogravimetric analysis, Karl Fischer analysis and <sup>1</sup>H NMR spectroscopy. The purity value is calculated as per Equation 1.

Purity =  $(100 \% - I_{ORG}) \times (100 \% - I_{VOL} - I_{NVR})$ 

Equation 1

I<sub>ORG</sub> = Organic impurities of related structure, I<sub>VOL</sub> = volatile impurities, I<sub>NVR</sub> = non-volatile residue

The purity estimate by qNMR was obtained using a combination of the three proton doublet at 1.04 ppm, the three proton singlet at 2.70 ppm and the one proton quartet at 3.47 ppm against a certified internal standard of maleic acid.

Supporting evidence is provided by headspace GC-MS analysis of occluded solvents and elemental microanalysis.

GC-FID:	Instrument: Column: Program: Injector: Detector Temp: Carrier: Split ratio: Relative mass fraction of Initial analysis:	Varian CP-3800/Agilent 7890 VF-1MS, $30.0 \text{ m} \times 0.32 \text{ mm} \text{ I.D.} \times 0.25 \text{ µm}$ 100  °C (2  min), 10  °C/min  to  200  °C, 20  °C/min  to  300  °C (5  min) 250  °C 320  °C Helium 20/1 of main component: Mean = 99.9%, s = 0.03% (10 sub samples in duplicate, March 2012)
	Re-analysis: Re-analysis: Re-analysis: Re-analysis:	Mean = 99.9%, s = 0.02% (5 sub samples in duplicate, February 2013) Mean = 99.9%, s = 0.02% (5 sub samples in duplicate, February 2016) Mean = 99.9%, s = 0.01% (5 sub samples in duplicate, October 2018) Mean = 99.9%, s = 0.05% (5 sub samples in duplicate, August 2021)
GC-FID:	Instrument: Column: Program: Injector: Detector Temp: Carrier: Split ratio: Relative mass fraction of Initial analysis:	Varian CP-3800 HP-5, 30 m $\times$ 0.32 mm l.D. $\times$ 0.25 µm 100 °C (1 min), 5 °C/min to 170 °C, 20 °C/min to 300 °C (3 min) 250 °C 320 °C Helium 20/1 of main component: Mean = 99.9%, s = 0.03% (10 sub samples in duplicate, March 2012)
Thermogravimetric analysis:		Volatile content < 0.1% and non volatile residue < 0.2% mass fraction (March 2012)
Karl Fischer analysis:		Moisture content < 0.1 % mass fraction (March 2012, January 2013, December 2015 & August 2021) Moisture content = 0.14% mass fraction (November 2018)
QNMR:	Instrument: Field strength: Solvent: Internal standard: Initial analysis: Initial analysis: Initial analysis:	Bruker Avance-III-400 400 MHz $D_2O$ (4.79 ppm) Maleic acid (98.7% mass fraction) Mean (1.04 ppm) = 99.9%, s = 0.4% (5 sub samples, March 2012) Mean (2.70 ppm) = 100.0%, s = 0.4% (5 sub samples, March 2012) Mean (3.47 ppm) = 99.9%, s = 0.3% (5 sub samples, March 2012)

### Spectroscopic and other characterisation data

GC-MS:		Agilent 6890/5973 TG-1MS, 30 m x 0.25 mm l.D. x 0.25 $\mu$ m 90 °C (1 min), 10 °C/min to 180 °C (7 min), 30 °C/min to 300 °C (3 min) 250 °C 280 °C Helium, 1.0 mL/min 20/1 e free base is reported along with the major peaks in the mass spectrum. The latter are peratios and (in brackets) as a percentage relative to the base peak. 146 (4), 118 (4), 117 (7), 106 (4), 77 (11), 58 (100), 56 (6), 42 (5) m/z	
ESI -MS:	Instrument: Operation: Ionisation: EM voltage: Cone voltage: Peak:	Micromass Quatro LC Micro Positive ion mode, direct infusion at 20 μL/min ESI spray voltage at 3.5 kV positive ion 650 V 5 V 166.2 (M+H <sup>+</sup> ) <i>m/z</i>	
HS-GC-MS:	Instrument: Column: Program: Injector: Transfer line temp: Carrier: Split ratio: Solvents detected:	Agilent 6890/5973/G1888 DB-624, 30 m x 0.25 mm l.D. x 1.4 μm 50 °C (5 min), 7 °C/min to 120 °C, 15 °C/min to 220 °C (8.3 min) 150 °C 280 °C Helium, 1.2 mL/min 50/1 None	
TLC:	Conditions:	eselgel 60F254. Diethylamine/Hexane/ethyl acetate/methanol (1/10/20/10) ingle spot observed, $R_f = 0.16$ . Visualisation with UV at 254 nm	
IR:	Instrument: Range: Peaks:	Biorad FTS3000MX FT-IR 4000-400 cm <sup>-1</sup> , KBr powder 3274, 3012, 2937, 2866, 2731, 2475, 2435, 2038, 1589, 1457, 1374, 1039, 1006, 919, 763, 703, 634 cm <sup>-1</sup>	
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance-400 400 MHz MeOH-d <sub>4</sub> (3.31 ppm) $\delta$ 1.10 (3H, d, <i>J</i> = 6.7 Hz), 2.73 (3H, s), 3.37 (1H, m), 4.56 (1H, d, <i>J</i> = 9.2 Hz), 7.35-7.45 (5H, m) ppm	
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Gyro-300 75 MHz MeOH-d₄ (49.0 ppm) δ 12.6, 30.5, 61.7, 75.5, 128.2, 129.7, 129.8, 141.9 ppm	
Melting point:		183-185 °C	
Microanalysis:	Found: Calculated:	C = 59.9%; H = 8.1%; N = 7.0%; Cl = 17.3% (March, 2012) C = 59.6%; H = 8.0%; N = 6.9%; Cl = 17.6% (Calculated for $C_{10}H_{15}NO.HCl$ )	