Australian Government

Department of Industry, Science and Resources National Measurement Institute



# REFERENCE MATERIAL PRODUCT INFORMATION SHEET

## **NMIA D609**: 5α–Dihydrotestosterone sulfate anion (nominally NEt3 salt)

Report ID: D609.2020.03 (Ampouled 101109)

Chemical Formula: C25H45NO5S

Molecular Weight: 471.7 g/mol

Property value

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Batch No.	CAS No.	Mass per ampoule
97-000703	1633766-47-8	776 ± 48 μg

### IUPAC name: Triethylammonium $(5\alpha, 17\beta)$ -3-Oxoandrostan-17-yl sulfate

**Expiration of certification:** The property values are valid till 18 September 2025, i.e. five years from the date of re-certification provided the **unopened** material is handled and stored in accordance with the recommendations below. The material as issued in the unopened container and stored as recommended below should be suitable for use beyond this date, subject to confirmation of batch stability from the issuing body. The expiry date/shelf life does not apply to ampoules that have been opened. In such cases it is recommended that the end-user conduct their own in-house stability trials.

**Description:** The compound is supplied as a dried aliquot in a sealed ampoule under an atmosphere of argon. The reference material is intended for a single use to prepare a standard solution containing D609. This material was prepared by synthesis, and certified for identity and purity by NMIA.

Intended use: This reference material should be used for qualitative analysis only.

**Instructions for use:** Open the ampoule and carefully rinse the interior at least three times with a suitable organic solvent (e.g. acetonitrile/water). This will transfer 776  $\mu$ g of anhydrous 5 $\alpha$ -dihydrotestosterone sulfate anion. The mass of analyte in each ampoule is calculated from the assigned purity of the bulk and the concentration of bulk material in a stock solution used to prepare the ampoules.

Recommended storage: When not in use, this material should be stored at or below 4 °C in a closed container in a dry, dark area.

Stability: The long-term stability of the compound in solution has not been examined.

**Homogeneity assessment:** The homogeneity of the material was assessed using purity assay by HPLC with ELS detection on seven randomly selected ampoules of the material. The material was judged to be sufficiently homogeneous at this level of sampling as the variation in analysis results between samples was not significantly different at a 95% confidence level from that observed on repeat analysis of the same sample.

**Safety:** Treat as hazardous substance. Use appropriate work practices when handling to avoid skin or eye contact, ingestion or inhalation of dust. Refer to the provided safety data sheet.

S.R. Davies

Dr Stephen R. Davies, Team Leader, Chemical Reference Materials, NMI. 2 November 2022

This report supersedes any issued prior to 2 November 2022.

NATA Accreditation No. 198 / Corporate Site No. 14214.

Legal notice: Terms and Conditions associated with the provision of this reference material can be found on the NMIA website.

#### **Characterisation Report:**

HPLC:	Instrument: Column:	Waters Model 1525 Binary pump, 717 plus autosampler Alltech Alltima C-18 5 $\mu$ m (4.6mm × 150 mm)	
	Column oven:	55 °C	
	Mobile Phase:	A = MilliQ water; B = Acetonitrile	
		0-6 min 35% B; 6-13 min 35-70% B; 13-20 min 70%B; 20-25 min 70-35%B.	
		Both A and B were buffered with 0.05% Trifluoroacetic acid	
	Flow rate:	0.8 mL/min (Gradient)	
	Detector:	Waters ELSD 2424	
R	Relative peak area of the main component:		
	Initial analysis:	Mean = 99.6%, s = 0.01% (7 ampoules in duplicate, December 2010)	
	Re-analysis:	Mean = 99.7%, s = 0.01% (5 ampoules in duplicate, October 2012)	
	Re-analysis:	Mean = 99.7%, s = 0.01% (5 ampoules in duplicate, September 2015)	
	Re-analysis:	Mean = 99.6%, s = 0.01% (5 ampoules in duplicate, September 2020)	

#### The following analytical data was obtained on the bulk material subsequently used in the preparation of the ampoules.

#### **Characterisation Report:**

The identity was confirmed by a range of spectroscopic techniques, NMR, IR and MS. The purity value was obtained by mass balance from a combination of traditional analytical techniques, including HPLC with ELS detection, thermogravimetric analysis, Karl Fischer analysis and <sup>1</sup>H NMR spectroscopy. The purity value is calculated as per Equation 1

Purity =  $(100 \% - I_{ORG}) \times (100 \% - I_{VOL} - I_{NVR})$  Equation 1

 $I_{ORG}$  = Organic impurities of related structure,  $I_{VOL}$  = volatile impurities,  $I_{NVR}$  = non-volatile residue.

Supporting evidence is provided by elemental microanalysis.

HPLC:	Instrument: Column: Column oven: Mobile Phase:	Waters Model 1525 Binary pump, 717 plus autosampler Alltech Alltima C-18 5 $\mu$ m (4.6mm × 150 mm) 55 °C A = MilliQ water; B = Acetonitrile	
	Flow rate: Detector:	0-6 min 35% B; 6-13 min 35-70% B; 13-20 min 70%B; 20-25 min 70-35%B. Both A and B were buffered with 0.05% Trifluoroacetic acid 0.8 mL/min (Gradient) Waters ELSD 2424	
	Relative peak area of the main component:		
	Initial analysis: Re- analysis: Re- analysis: Re- analysis:	Mean = 98.1%, s = 0.2% (3 sub samples in duplicate, December 2000) Mean = 99.1%, s = 0.07% (5 sub samples in duplicate, November 2007) Mean = 99.2%, s = 0.09% (5 sub samples in duplicate, November 2008) Mean = 99.6%, s = 0.01% (7 sub samples in duplicate, November 2010)	
Karl Fischer analysis:		Moisture content 1.2% mass fraction. (2 sub samples, November 2007) Moisture content 0.95% mass fraction (2 sub samples, December 2010)	
Thermogravi	metric analysis:	Volatile content 0.6% mass fraction (November 2007) Non volatile residue was not determined	

#### Spectroscopic and other characterisation data

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ESI-MS:	Instrument: Operation: Ionisation: Peak:	Micromass Quatro LC Micro Negative ion mode, direct infusion ESI spray voltage at 4.5 kV negative ion 369 (MSO <sub>3</sub> ) <sup>-</sup> $m/z$
TLC:	Conditions:	Kieselgel 60F <sub>254</sub> . Chloroform/methanol/water (70:20:2) Single spot observed, $R_f = 0.3$
IR:	Instrument: Range: Peaks:	Bruker Alpha Platinum ATR 4000-400 cm <sup>-1</sup> , neat 3500, 2740, 2679, 2492, 1719, 1234, 1063, 1004, 856, 804 cm <sup>-1</sup>
<sup>1</sup> H NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker Avance III-500 500 MHz Acetone- $d_6$ (2.05 ppm) $\delta$ 0.65 (3H, s), 0.95 (3H, s), 1.16 (9H, t, $J = 7.3$ Hz), 3.09 (6H, q, $J = 7.3$ Hz) 3.95 (1H, m) ppm <sup>1</sup> H NMR (6/12/2007) run under QNMR conditions (no internal standard) indicates that only 92% of the 5 $\alpha$ -dihydrotestosterone sulfate anion is associated with the triethylammonium ion. The remaining 8% is protonated.
<sup>13</sup> C NMR:	Instrument: Field strength: Solvent: Spectral data:	Bruker DMX-500 126 MHz Acetone- <i>d</i> <sub>6</sub> (29.8 ppm) δ 9.0, 11.4, 12.1, 20.9, 23.4, 28.5, 28.7, 31.2, 35.2, 35.7, 37.0, 38.0, 38.3, 42.6, 44.5, 46.2, 46.4, 50.4, 53.6, 210.8 ppm
Microanalysis:	Found: Calculated:	C = 63.0%; H = 9.5%; N = 2.8% (August 1999) C = 63.7%; H = 9.6%; N = 3.0% (Calculated for $C_{25}H_{45}NO_5S$ )